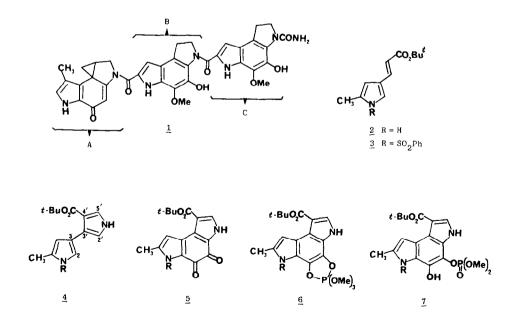
STUDIES ON THE ANTITUMOR AGENT CC-1065

Regiospecific Introduction of the Oxygen Functionality for the Synthesis of the B/C Component of CC-1065.

Philip Magnus* and Serge Halazy Department of Chemistry, Indiana University, Bloomington, Indiana 47405

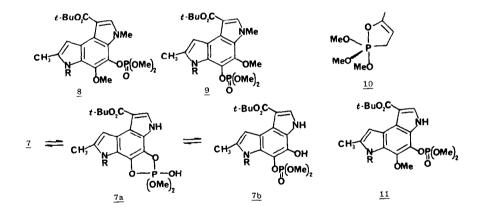
SUMMARY: The $3,3^1$ -bipyrrole <u>4</u> is readily converted into <u>o</u>quinone <u>5</u>, which was regiospecifically elaborated into the CC-1065 \overline{B}/C -model component <u>15</u> by sequential reduction, alkylation and reduction reactions.

Recently, we have reported the successful application of the $3,3^{1}$ bipyrrole strategy for synthesizing the <u>A</u>-portion of CC-1065 <u>1</u>.¹ Here we describe the use of this strategy for the construction of the key features of the B/C components.



Treatment of <u>t</u>-butyl 2,4-pentadienoate² with MeCH(Ts) NC/NaH/THF/20°C gave the 1,6-addition product <u>2</u> (75%), which was directly converted into the <u>N</u>-phenylsulfonyl derivative <u>3</u> (65%) by standard methods. Pyrrole annulation of <u>3</u> using TOSMIC/NaH/THF/20°C resulted in the 3,3¹-bipyrrole <u>4</u> (75%). The purpose of the -CO₂Bu^t group is to block electrophilic substitution in the adjacent 5¹-position, and to provide the necessary electrophilicity needed to make <u>2</u> and <u>4</u>.

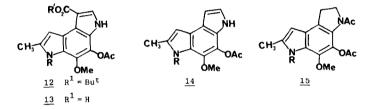
When a solution of the bipyrrole $\underline{4}$ in CH_2Cl_2 at $-78^{\circ}C$ was treated with oxalyl chloride³ in ether, and the mixture warmed to $20^{\circ}C$, the <u>o</u>-quinone <u>5</u> crystallized (85%), λ_{max}^{MeOH} 365, 255 and 220nm (ε , 3800, 4500 and 7500). Reduction of the <u>o</u>-quinone <u>5</u> using $P(OMe)_3^{*}$ in benzene gave the cyclic oxyphosphorane <u>6</u> (75%), which was rapidly hydrolyzed in wet THF to a single phenolic phosphate ester <u>7</u> (100%), m.p. 169-172°C (benzene/hexane). Treatment of <u>7</u> with $Me_2SO_4/acetone/K_2CO_3$ gave <u>8</u> (70%) and <u>9</u> (8%). Presumably, <u>7</u> can equilibrate via <u>7a</u> to <u>7b</u>, and thus give rise to the undesired regioisomer <u>9</u>. The structure of 8 was established by single crystal X-ray crystallography.⁵



The <u>N</u>-methylation of <u>7</u> was avoided by treatment of <u>7</u> with the cyclic oxaphospholene <u>10</u> [prepared from P(OMe)₃ and methylvinylketone]⁶ in THF/20°C (neutral conditions) to give <u>11</u> (50%), m.p. 102-103°C. The structure of <u>11</u> was established by <u>N</u>-methylation, NaH/MeI, to give <u>8</u>.

Removal of the phosphate group with NaOMe/MeOH, followed by Ac2O gave

<u>12</u> (80%). When a solution of <u>12</u> in CHCl³ was treated with trifluoroacetic acid, at 20°C, the acid <u>13</u> was formed. While <u>13</u> proved to be resistant to the usual decarboxylation conditions associated with indole and pyrrole 3-carboxylic acids, triethylamine in toluene at 110°C/6h. gave <u>14</u> (80%). The unsubstituted 2,3-double bond in <u>14</u> was selectively reduced by exposure of <u>14</u> to Et₃SiH/TFA/0°C,⁷ followed by Ac₂O to give the diacetate <u>15</u> (60%).



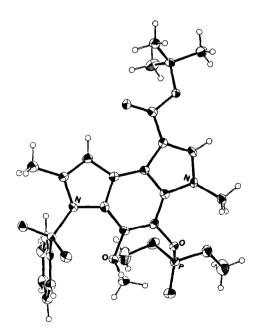
In summary, the $3,3^1$ -bipyrrole strategy, combined with <u>o</u>-quinone formation, $P(OMe)_3$ reduction, and oxaphospholene <u>O</u>-methylation provides a practical solution for constructing the requisite oxygenation pattern in the B/C portion of CC-1065.⁸

ACKNOWLEDGEMENTS: The National Institute of Health, grant number GM 32718, is thanked for financial support. Dr. John Huffman, Molecular Structure Center, Indiana University, Bloomington, Indiana 47405, is thanked for the X-ray crystallographic structural determination of 8.

References and Footnotes

 P. Magnus and T. Gallagher, J.C.S. Chem. Comm., 389 (1984). For other work in this area see: W. Wierenga, J. Am. Chem. Soc., 5621, <u>103</u> (1981); G.A. Kraus, S. Yue, and J. Sy, <u>J. Org. Chem.</u>, 283, <u>50</u> (1985); V.H. Rawal and M.P. Cava, <u>J.C.S. Chem. Comm.</u>, 1526 (1984); G.A. Kraus and S. Yue, <u>Ibid</u>., 1198 (1983); R.J. Sundberg and T. Nishiguchi, <u>Tetrahedron Lett</u>., 4773 (1983).

- I.E. Muskat, B.C. Becker, and J.S. Lowenstein, <u>J. Am. Chem. Soc.</u>, 326, <u>52</u>, (1930); J.P. Idoux and H. Ghane, <u>J. Chem. Eng. data</u>, 157, <u>24</u> (1979).
- 3. A. Treibs and F.H. Kreuzer, <u>Liebigs. Ann. Chem</u>., 105, <u>721</u> (1969); J.L. Archibald and M.E. Freed, J. Het. Chem., 335 (1967).
- 4. F. Ramirez and N.B. Desai, <u>J. Am. Chem. Soc</u>., 2652, <u>82</u> (1960).
- 5. The structure of <u>8</u> has been determined crystallographically see ortep below. Details are available on microfiche, from the Indiana University Chemistry Library - request Structure Report No. 83128.
- 6. W.G. Voniken and H.M. Buck, <u>Rec. Trav. Chim</u>., 210, <u>93</u>, (1974).
- 7. G. Guillerm, F. Frappier, J.C. Tabet, and A. Marquet, <u>J. Org. Chem</u>., 3776, 42 (1977).
- 8. All new compounds were characterized by ¹H NMR, IR and micro-analytical/ MS data.



ORTEP DRAWING OF 8